

Chapter 10 Chemical Quantities Worksheet Answers

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Chapter 10 Chemical Quantities Worksheet

Section 10.1 □ The Mole: A Measurement of Matter. You often measure the amount of something by count, by mass, or by volume. A mole (mol) of a substance is 6.02×10^{23} representative particles of that substance. 6.02×10^{23} is called Avogadro's number. 1 mole = 6.02×10^{23} representative particles

Chapter 10 □ Chemical Quantities

CHEMICAL QUANTITIES SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287□296) This section defines the mole and explains how the mole is used to measure matter. It also teaches you how to calculate the mass of a mole of any substance. Measuring Matter (pages 287□289)

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SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287□296)

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Chemical Quantities THE MOLE AND QUANTIFYING MATTER 10.1 The Mole: A Measurement of Matter Essential Understanding The mole represents a large number of very small particles. Reading Strategy Frayer Model The Frayer Model is a vocabulary development tool. The center of the

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