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Part I. Stoichiometry 1.  $1 \text{ ZnI}_2 \rightarrow 1 \text{ Zn} + 1 \text{ I}_2$   
How many grams of iodine will you produce if you begin your rxn with 56.7g of zinc iodide? 56.7 g 1 mole 1

# Online Library Chapter 12 Stoichiometry Vocabulary

Review  
mole 253.8 g 319.19 g 1 mole 1 mole  
= 45.08 grams of Iodine 2. 1  
 $\text{Pb}_2(\text{CO}_3)_3 + 6 \text{K} \rightarrow 2 \text{Pb} + 3 \text{K}_2\text{CO}_3$

Chapter 12 REVIEW: Stoichiometry,  
Theoretical, Actual ...

STOICHIOMETRY 12 Vocabulary

Review Match the correct vocabulary

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Review  
term to each numbered statement.  
Write the letter Of the correct term on.  
the line. Column A 1. the starting  
materials in a chemical reaction 2. a  
conversion factor derived from the  
coefficients of a balanced chemical  
equation interpreted in terms Of moles

# Online Library Chapter 12 Stoichiometry Vocabulary

Mister Chemistry Welcomes You! □

Chemistry teacher at ...

Chemistry Chapter 12 vocabulary.

Stoichiometry. STUDY. PLAY. Mole.

$6.02 \times 10^{23}$  representative particles.

Reactants. A starting material in a

chemical reaction. Products. The

elements or compounds produced by



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Review a chemical reaction. Limiting reactant.

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The calculations of quantities in chemical reactions in a subject of chemistry

- Mole ratio- a conversion factor derived from the coefficients of a balanced chemical equation interpreted in terms of moles

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react 25.0 liters of oxygen gas at STP

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with 50.0 ml of ethanol liquid. Ethanol has a density of 0.789 g/ml. Use this reaction to solve the following problems.

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