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SHORT ANSWER Answer the following questions in the space provided. 1. d How many quantum numbers are used to describe the properties of electrons in atomic orbitals? (a) 1 (c) 3 (b) 2 (d) 4 . 2. a A spherical electron cloud surrounding an atomic nucleus would best represent

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atomic orbital but only if the electrons have opposite spins
Hund's rule states that single electrons with the same spin must occupy each equal-energy orbital before additional electrons with opposite spins can occupy the same orbitals

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