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Atoms. Protons, neutrons, and electrons. Nucleus. The small, positively charged center of the atom.

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Periodic Table Parts of an Atom • An  
atom consists of a nucleus  
surrounded by one or more electrons  
• Atoms are electrically neutral w/  
the same number of protons as  
electrons. • Majority of the atom is  
empty space. If nucleus were the size

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of a pencil eraser, the closest electron would be 100 yards away!

## Chapter 16

BeccaSwan1. Physical Science

Chapter 16 - Properties of Atoms and the Periodic Table. atom. nucleus.

proton. neutron. Smallest unit of matter that retains the properties of an elem.... A region that is located at the center of an atom and contains....

A subatomic particle that has a positive charge and that is fo....

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Liquid Figure 4 The particles in a liquid are moving more freely than the particles in a solid. They have enough kinetic energy to slip out of the ordered arrangement of a solid.  
Figure 5 In gases, the particles are far apart and the attractive forces between the particles are overcome. Gases do not have a defi-

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Table. atom. nucleus. proton.

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neutron. Smallest unit of matter that retains the properties of an elem.... A region that is located at the center of an atom and contains.... A subatomic particle that has a positive charge and that is fo....

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